## SYNERGY SCHOOL OF ENGINEERING LESSON PLAN FOR APPLIED CHEMISTRY(2<sup>ND</sup>SEM)

Discipline :	Semester: 2 <sup>ND</sup>		Name of teaching faculty Ms Amrita A. Nanda/ Ms J. Behera
Subject: APPLIED CHEMISTRY	No's of days/ per week class allotted :4		Semester start from: 04/02/25 - 17/05/25
WEEK	CLASS	UNIT	TOPICS
1	1	1	Rutherford model o atom -Postulates,
	2	1	Bohr's theory- Postulates
	3	1	Hydrogen spectrum explanation based on Bohr's model of atom, Heisenberg uncertainty principle
	4	1	Quantum numbers – orbital concept. Shapes of s,p and d orbitals
2	1	1	Pauli's exclusion principle, Hund's rule of maximum multiplicity  Ausbau rule
	1	1	El designation
	3	1	Concept of chemical bonding – cause of chemical bonding, types of bonds
		1	Landa handing with examples
	1	<del></del>	Covalent bond (H <sub>2</sub> , F <sub>2</sub> , HF hybridization in BeCl <sub>2</sub> , BF <sub>3</sub> , CH <sub>4</sub> , NH <sub>3</sub> ,
3	1		11.0)
	2	1	Coordination bond in NH <sub>4</sub> , and anomalous properties of NH <sub>3</sub> and
	3	1	Solute, solvent and solution, methods to express the concentration
	4	1	Dam mass percentage volume percentage and mole fraction
	<del></del>	2	Graphical presentation of water distribution on Earth; introduction
4	2	2	Soft and hard water : Cause of poor lathering of soap in hard water,
	3	2	Salts causing water hardness, unit and simple numericals
		2	Problems caused by the use of hard water in boiler (scale and
	4		Quantitative measurement of water hardness by ETDA method.
5	1	2	TDS by alkalinity
		- 2	Water softening techniques – soda lime process, zeolite process
	4	2	Water softening techniques- ion exchange and Municipal water treatment sedimentation,
		2	Congulation filtration sterilization
6	2 .	2	Water for human consumption enlist Indian standard specification of drinking water (collect data and understand standards)
	3	3	Natural occurrence of metals – minerals
	- 1	3	Ores of iron, aluminium and copper,
		$-\frac{3}{3}$	Gangue (matrix), flux, slag,
7	1	3	General principles of metallurgy
	3	3	Extraction of - iron from haematite ore using blast furnace
		3	Extraction of aluminium from bauxite
8	1	3	Alloys – definition, purposes and types with examples, properties
	2		General chemical composition, composition based applications

	3	3	Port land cement and hardening
	4	3	Glasses Refractory and Composite materials.
9	1	3	Monomer, homo and co polymers, degree of polymerization;
	26.0.00		thermoplastics and thermosetting plastics
	2	3	PVC, PS, PTFE, nylon – 6,
	3	3	Nylon-6,6 and Bakelite; rubber and vulcanization of rubber
	4	4	Definition of fuel and combustion of fuel, classification of fuels,
10	1	4	Calorific values calculation of HCV and LCV using Dulong's formula
	2	4	Proximate analysis of coal solid fuel petrol and diesel
	3	4	Fuel rating (octane and cetane numbers)
	4	4	Chemical composition, calorific values and applications of LPG,
	-		CNG water gas
11	1	4	Chemical composition, calorific values and applications of Coal
	V. 4 5 *	1	gas producer gas and biogas.
	2	4	Lubrication – function and characteristic properties of good
	_	-	lubricant
	3	4	Classification with examples, lubrication mechanism –
	3		bydrodynamic and boundary lubrication,
	4	4	Physical properties (viscosity and viscosity index, oiliness, flash
	Section 1		and fire point).
12	1.	4	chemical properties (coke number, total acid number
1.2			saponification value)
	2	5	Oxidation, reduction and redox reactions.
			Electrolytes and non-electrolytes
	3	5	Faradays laws of
			electrolysis and simple numerical problems
	4	5	Industrial Application of Electrolysis –Electrometallurgy
		(78	electroplating and electrolytic refining.
13	1	5	Primary cells – dry cell,
	2	5	Secondary cell - commercially used lead storage battery, fuel and
			Solar cells.
	3	5	Definition, types of corrosion (chemical and electrochemical)
	4	5	H <sub>2</sub> liberation and O <sub>2</sub> absorption mechanism of electrochemical
			corrosion,
14	1	5	Factors affecting rate of corrosion
	2	5	Internal corrosion preventive measures -Purification, alloying and
			heat treatment and
	3	5	External corrosion preventive measures: metal (anodic, cathodic)
			coatings,
	4	5	External corrosion preventive measures -organic
			inhibitors.
15	1		Revision
	2		Revision
	3		Revision
	1		Revision

